

- a -

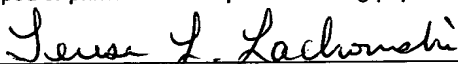
APPLICATION FOR UNITED STATES PATENT

A METHOD FOR CONTROLLING DEPOSIT FORMATION IN
GASOLINE DIRECT INJECTION ENGINE BY USE OF A FUEL
HAVING PARTICULAR COMPOSITIONAL CHARACTERISTICS

Applicants: Noyes L. Avery
Paul P. Wells

CROSS REFERENCE TO RELATED APPLICATION:

This application claims the benefit of U.S. Provisional
application 60/242,814 filed October 24, 2000,

"EXPRESS MAIL" mailing label	
number	EF312156153US
Date of Deposit	September 28, 2001
I hereby certify that this paper or fee is being deposited with the United States Postal Service "Express Mail Post Office to Addressee" service under 37 CFR 1.10 on the date indicated above and is addressed to the Commissioner for Patents, Washington, D.C. 20231.	
Teresa L. Lachowski	
(Typed or printed name of person mailing paper or fee)	
	
(Signature of person mailing paper or fee)	

Case No. JJA-0106

A METHOD FOR CONTROLLING DEPOSIT FORMATION IN
GASOLINE DIRECT INJECTION ENGINE BY USE OF A FUEL
HAVING PARTICULAR COMPOSITIONAL CHARACTERISTICS

5 $\frac{1}{2} \text{ IN } \frac{1}{2}$
BACKGROUND OF THE INVENTION

FIELD OF THE INVENTION

10 [0001] The present invention relates to a method for operating a spark
ignition, direct injection gasoline engine to control the formation of deposits on
the fuel injectors by combustion in said engine of a fuel characterized as a
gasoline of particular composition.

DESCRIPTION OF THE RELATED ART

15 [0002] Gasoline direct injection (GDI) engine technology has been
investigated for about 80 years, but injector coking is still a major concern.
Because gasoline direct injection engines have a fuel economy benefit of
15-30% and result in the production of lower levels of greenhouse gas CO₂
20 emissions and exhibit a power improvement of 5-15% per unit volume of engine
displacement, they have continued to be investigated and developed despite the
technical challenges of fuel management control, engine deposits, exhaust
emissions control and injector fouling. An especially attractive feature of
gasoline direct injection engines is the reduction in the octane requirement of the
25 engine. Because of in-cylinder fuel evaporation and resulting charge cooling
effect, up to a six number lower octane requirement can be exhibited by a GDI.
However, as previously stated, and despite the advantages demonstrated by GDI
engines, early GDI engines such as the Texaco TCP of 1951 and the Ford
PROCOS of 1968 suffered from severe deposit problems which contributed to
30 the demise of those programs.

09967265-082801

[0003] Gasoline direct injector engines, however, have continued to be developed in Japan and Europe. Mitsubishi, Toyota, Nissan, Honda, Mazda, Renault and VW have all indicated a commitment to pursuing GDI engine technology and commercializing vehicles powered by GDI engines.

[0004] However, injector fouling will continue to be an area of concern for manufacture and vehicle owner.

10 DESCRIPTION OF THE FIGURES

[0005] Figure 1 reports the percent flow loss versus tip temperature for the eleven fuels evaluated.

15 [0006] Figure 2 reports the percent flow loss versus tip temperature for three fuels of controlled aromatics content and Howell EEE reference fuel.

DESCRIPTION OF THE INVENTION

20 [0007] Pump gasoline can contain anywhere from about 12 to about 65 vol% aromatics, typically about 20-40 vol%, more typically about 25 to 35 vol% aromatics. (See "How Gasoline Has Changed", L. M. Gibbs, SAE Technical Paper Series #932828, October 1993.) The present invention is directed to the discovery that the nature of the aromatics is important in controlling the formation of injector deposits in gasoline direct injection (GDI) engines.

[0008] It has been discovered that injector deposit formation in a GDI engine is controlled by combusting in the engine a fuel comprising gasoline containing from about 12 to about 65 vol % aromatics wherein the source of the aromatics is

a stream selected from the group consisting of reformate, fluid catalytic cracker stream and mixtures thereof, and wherein with respect to the fluid catalytic cracker stream, light fluid cat cracker stream constitutes about 70% to 100% of the fluid cat cracker stream, preferably about 85% to 100%, most preferably about 95% to 100% of the total FCC stream. It is preferred that the aromatics source is reformate.

[0009] Preferably it has been discovered that in the operation of a GDI engine wherein injector deposit formation is controlled by combustion in the engine of a fuel comprising gasoline characterized by having a T₉₀ in the range of about 150-182°C, preferably about 160-182°C, an olefins content in the range of about 3.6-20 vol%, preferably about 3.6 to 10 vol%, more preferably less than 5 vol%, a sulfur content in the range of about 50-400 ppm, preferably about 150 ppm, an aromatics content in the range of about 10-45 vol%, preferably about 25 to 37 vol% injector deposit formation control is improved when the aromatics are secured from reformate, a fluid cat cracker (FCC) stream or mixture thereof, and wherein with respect to the fluid cat cracker stream, light FCC stream constitutes about 70% to 100%, preferably about 85% to 100%, most preferably about 95% to 100% of the total FCC stream, it being preferred that the aromatics source is reformate.

[0010] Preferably the fuel is unleaded gasoline and may contain quantities of other, typical gasoline additives such as detergents, including Mannich bases, polyisobutyl amines, polyether amines or succinimides, preferably Mannich bases, in an amount in the range 25 ptb to 200 ptb, antioxidants such as phenolic or aminic antioxidants, lubricity additives such as fatty acids or fatty acid esters, oxygenates such as alcohol or ethers, e.g., methyltertbutyl, ether, anti icing additives, demulsifiers, corrosion inhibitors, dyes, etc.

[0011] It has been discovered that the control and reduction in the formation of deposits on the injectors of spark ignition gasoline direct injection engines operating on the fuels characterized by the above recited T₉₀, sulfur content olefins content and aromatics content is further and unexpectedly improved when the source of aromatics in the fuel is selected from the group consisting of reformat, light fluid cat cracker streams or mixtures thereof, preferably reformat.

[0012] Catalytic cracking, or cat cracking as it is commonly referred to, is an established and widely used process in the petroleum refining industry for converting petroleum oils of relatively high boiling point to more valuable lower boiling products, including gasoline and middle distillates, such as kerosene, jet fuel and heating oil. The preeminent catalytic cracking process now in use is the fluid catalytic cracking process (FCC) in which a preheated feed is brought into contact with a hot cracking catalyst which is in the form of a fine powder, typically having a particle size range of from about 10-300 microns and with a mean particle size of about 70-100 microns, for the desired cracking reaction to take place. The catalyst is fluidized by the hydrocarbon vapors. Catalysts which are conventionally used are based on zeolites, especially the large pore synthetic faujasites, zeolites X and Y. During the cracking, coke and hydrocarbonaceous material are deposited on the catalyst particles. This results in a loss of catalyst activity and selectivity. The coked catalyst particles and associated hydrocarbon material are subjected to a stripping process, usually with steam, to remove as much of the hydrocarbonaceous material as is technically and economically feasible. The striped catalyst particles, containing non-strippable coke, are removed from the stripper and sent to a regenerator where they are regenerated by contact with an oxygen-containing gas, typically air or a mixture of air and oxygen, at an elevated temperature. This results in the combustion of the coke which is a strongly exothermic reaction which, besides removing the coke,

0996755 5927966
serves to heat the catalyst to the temperatures appropriate for the endothermic cracking reaction. The process is carried out in an integrated unit which comprises cracking, stripping and regenerating zones and associated ancillary equipment. Typically the cracking and stripping zones are associated with a single vessel or unit, with the regenerator being a separate unit. The catalyst is continuously circulated from the reactor or reaction zone, to the stripper and then to the regenerator and back to the reactor. The catalyst circulation rate is typically adjusted relative to the feed rate of the oil to maintain a heat balanced operation in which the heat produced in the regenerator is sufficient for maintaining the cracking reaction, with the circulating regenerated catalyst being used as the heat transfer medium.

[0013] As used in the present specification and the appended claims, by light fluid cat cracking stream fraction is meant a fraction characterized as having an initial boiling point (IBP) in the range of about 90°F to 100°F, a T_{10} in the range of about 130°F to 150°F, a T_{90} in the range of about 280°F to 300°F, and a final boiling point in the range of about 330°F to 350°F, preferably an IBP in the range of about 90-95°F, a T_{10} in the range of about 130-140°F, a T_{90} in the range of about 280-290°F, and a FBP in the range of about 330-340°F, most preferably an IBP of about 90°F, a T_{10} of about 130°F, a T_{90} of about 280°F, and a FBP of about 330°F.

[0014] Reforming is the process whereby the higher boiling, but lower value, lower octane portion of gasoline, e.g., virgin naphtha be it straight run or produced by hydrocracking is converted into higher octane gasoline by the rearrangement of the molecular structure of the hydrocarbon. The primary physical change which occurs are dehydrogenation of cyclic compound (cycloparaffins) into aromatics, e.g., cyclohexane and naphthenes into benzene,

toluene, xylene, etc., while straight chain paraffins of sufficient carbon number (C_6 and greater can undergo dehydrocyclization for conversion similarly into benzene, toluene, ethyl benzene, xylene, etc.

5 **[0015]** Reforming can occur either thermally (thermal reforming) or via catalysis (catalytic reforming). In either case the end product obtained is an aromatics enriched stream suitable for addition to the mogas pool. Preferably the reforming is catalytic reforming as the end product obtained is much higher in octane than can be secured by thermal reforming.

10 **[0016]** Reformate suitable for use in the present invention is characterized by a RON of about 95 to 105, preferably about 98 to 102, an initial boiling point in the range of about 90 to 95°F, a T_{10} in the range of about 140 to 145°F, a T_{90} in the range of about 310-320°F, a final boiling point (FBP) in the range of about
15 400 to 430°F and an aromatics content of about 40-70%, preferably an IBP of about 95°F, a T_{10} of about 145°F, a T_{90} of about 310°F, an FBP of about 400°F, and an aromatics content of about 50% to 65%.

20 **[0017]** The amount of reformate, fluid cat cracker stream stock or mixture thereof, preferable reformate, included in the gasoline is an amount sufficient to produce in the final gasoline product an aromatics content within the aforesaid range of about 10 to 45 vol%, preferably about 20 to 40 vol%, more preferably about 25 to 37 vol%.

25 **[0018]** If a mixture of reformate and fluid cat cracking stream stock is employed the ratio of reformate to fluid cat cracker stream product is in the range of about 100:0 to 25:75, preferably 100:0 to 75:25, most preferably about 100:0 to 80:20. It is preferred, however, that the aromatics in the gasoline be attributable predominately to reformate, i.e., about 70% to 100% of the

aromatics come from reformat, preferably about 80 / to 100% of the aromatics come from reformat.

EXPERIMENTAL

5

[0019] An eleven-fuel test matrix was designed around T₉₀, sulfur level, and olefins level indicated in European Gasoline specifications for year 2000. Three additional fuels were blended to study the effects of fuel aromatics content. The fuels are base fuels and contained no detergents or other additives other than antioxidant. They were formulated using refinery stream blends to meet the specific compositional targets and to insure that such fuels could be produced commercially.

10

[0020] The engine test bed was a conventional dual sparkplug 2.2 liter Nissan engine modified for direct injection using one of the sparkplug holes. Injector tip temperatures were controlled to range from 120 to 184°C to match the injector tip temperatures typically encountered in the two main types of GDI designs: spray-guided and wall-guided combustion systems.

15

[0021] Spray-guided systems involve a controlling mounted injector located close to the spark plug. The distance of separation is such that the spark ignites the edge of the spray cone. Because of the close proximity of the injector tip to the combustion event, this type of system is prone to injector coking due to high tip temperatures.

20

25

[0022] Wall-guided systems have injectors mounted at an angle, and the spray is deflected off the top of the piston towards the spark plug. Because the injector tip allows lower tip temperature as a consequence of being a greater distance from the flame front and due to the greater air movement around the tip

reducing the amount of residual fuel at the tip available for combustion, this type of system experiences less injector coking.

[0023] The 2.2 liter engine was modified to run in a homogeneous direct injection mode. Modifications included replacing the exhaust side spark plug with pre-production high-pressure common-rail direct injectors, removing the original equipment manufacturer's spark and fuel system and installing a high pressure fuel system and universal engine controller. For homogeneous combustion, flat top pistons and the conventional gasoline spark ignition combustion chamber design were found to be sufficient. The injectors were located on the hot (exhaust) side of the engine to favor high tip temperature to favor injector deposit formation.

[0024] The test engine specifications are presented in detail in Table 1.

TABLE 1

Type	Four Cylinder In-Line 2.2 L Nissan Engine Converted for DI Operation
Displacement	2187 cubic centimeters
Plugs/Cylinder	1 (stock configuration: 2)
Valves/Cylinder	2
Bore	87 millimeters
Stroke	92 millimeters
Fuel System	Common Rail High Pressure Direct Injection
Fuel Pressure	6900 kPa (closed loop)
Engine Controller	Universal Laboratory System
Injection Timing	300 degrees BTDC
Ignition Timing	20 degrees BTDC
Coolant Temperature	85°C
Oil Temperature	95°C

[0025] It was found that the key operating parameters of the engine were inlet air and fuel temperature, engine speed and engine load.

5 [0026] The inlet air and fuel temperatures were controlled at 35°C and 32°C respectively.

[0027] At constant inlet air and fuel temperature and engine load, tip temperature remained constant at engine speeds of 1500, 2000, 2500 and 3000
10 rpm.

[0028] However, at constant engine speed and constant inlet air and fuel temperatures, tip temperature increased with load. For five load points of 200, 300, 400, 500 and 600 mg/stroke air charge, increasing tip temperature of 120,
15 140, 157, 173 and 184°C, respectively, were observed.

[0029] Based on this information test conditions were set at a constant engine speed of 2500 rpm, inlet air temperature of 35°C, inlet fuel temperature of 32°C with tip temperature being controlled by controlling the load. In the case of each
20 first tested, at least four load points were run for each fuel.

[0030] The test is divided into three periods: engine warm-up, an operator assisted period, and a test period.

25 [0031] Engine speed was controlled using the engine dynamometer controller, and the engine throttle was manipulated to control air charge using a standard automotive airflow meter as feedback in a closed-loop control system. Engine fueling was controlled in two ways. During warm-up, injector pulse width was controlled using a standard mass airflow strategy and exhaust gas

09067265-082801
T00280-592/9660

sensor controlling the air/fuel mixture to stoichiometric. During the operator interaction period, the pulse width was manually set for each injector using wide-range lambda sensors in the exhaust port of each cylinder. Fuel flow was measured using a volumetric flow meter and a temperature-corrected density value was used to calculate mass flow. Ignition timing was held constant at 20° BTDC throughout the test. Inlet air temperature was controlled to 35 +/- 2°C and fuel temperature at the inlet to the high-pressure pump was controlled to 32 +/- 2°C. Data were sampled ten times per second and averaged to form a record of all recorded parameters every ten seconds during the test.

10

[0032] Data acquisition began as soon as the engine was started. The engine idled for one minute before the speed was raised to 1500 rpm and the air charged (load) to 300 mg per stroke to warm the engine to operating temperature. During this 30-minute warm-up period coolant and oil temperatures were linearly raised from 40 to 85 +/- 2°C and 40 to 95 +/- 1 2°C, respectively.

15

[0033] At the end of warm-up, engine speed was increased to 2500 rpm, and the air charge adjusted to the test load target, which ranged from 100 to 600 mg air/stroke depending on the desired injector tip temperature. Within five minutes injector pulse width for each cylinder was manually adjusted to a lambda target value of 0.800 +/- 0.005.

20

[0034] For the remainder of the test, pulse width, speed, and air charge remained constant. The change in fuel flow for the engine and the calculated change in fuel flow, based on lambda of each individual cylinder, were the measure of the injector flow decrease due to deposit formation.

25

[0035] Each fuel was run at four to five load conditions as previously stated. The points were repeated for each fuel using the average injector flow loss for

the engine as a function of injector temperature to form a characteristic curve for each fuel. Injector deposit formation was followed by measuring total engine fuel flow at fixed speed, air charge (mass of air per intake stroke), and the lambda signal from each cylinder over a test period of six hours.

5

[0036] To help minimize injector-to-injector variability the same set of injectors was used for all tests at a particular engine load, with each injector always in the same cylinder. Different sets of injectors, however, were used for different load conditions.

10

[0037] After each test, injector tips were photographed and external deposits scraped off and collected together as one for each fuel. Analyses by scanning electron microscope (SEM), and infrared spectroscopy (IR) were then conducted. The internal deposits were flushed out in a special rig with 200 ml pentane and a mixture of MTBE/pentane/methanol in a ratio of 1/0.5/0.5 with the injector powered. The solvent mixture was evaporated, and the residue analyzed by SEM and IR. For comparison, each fuel was similarly concentrated and the residue analyzed as above. A sample of unused lubricant was similarly analyzed. This was done to define relative contributions of fuel and lubricant to the deposit.

15
20

[0038] An injector-cleaning rig was equipped with a high-pressure fuel rail and a pump that cycled cleaning fluid through the injectors. Injectors were pulsed to allow the cleaning fluid through. This cleaning process brought the injectors to their baseline flow conditions as confirmed by the repeatability of numerous reference fuel testing during the study.

25

T03230"99249660

FUELS

[0039] The fuels matrix was designed to examine fuel composition effects on deposit formation, with maximum values of T₉₀, sulfur, and olefins based on European fuel specifications in the year 2000 (olefins: 18 vol% maximum; sulfur: 150 ppm maximum; E₁₀₀: 46% minimum; E₁₅₀: 75% minimum). The test fuels were blended from refinery streams to match the desired test matrix design in Table 2. Howell EEE gasoline was included as the eleventh fuel. None of the fuels were additized.

TABLE 2 - Specifications gasoline matrix, test fuels

Fuel Number	T ₉₀ (°C)	Olefins (vol%)	Sulfur (ppm)	Aromatics (vol%)
1	160	5	30	16
2	182	5	30	16.4
3	160	20	30	18.0
4	182	20	30	22.8
5	160	5	150	20.6
6	182	5	150	31.9
7	160	20	150	10.4
8	182	20	150	20.8
9	171	12.5	90	19.9
10	171	12.5	400	29.8
Howell EEE	160	1.2	20	26.6

15 TEMPERATURE EFFECT ON FLOW LOSS

[0040] For most fuels tested flow loss becomes increasingly severe with increasing tip temperature (load) up to a maximum tip temperature of 173°C

(load 500 mg/stroke air), but surprisingly improves slightly at even higher tip temperature of 184°C (load 600 mg/stroke air) (Figure 1). The observed reduction of deposits at the highest injector tip temperature may be related to the reported temperature effects on combustion chamber deposits (CCD), where an inverse relationship between wall temperature and CCD was determined.

FUEL COMPONENT EFFECT

[0041] Figure 1 shows the injector plugging characteristics of all fuels tested at various tip temperatures. There is a large difference between these fuels with the most significant occurring at tip temperature of 173°C. Four replicate tests of Fuel 7 at 173°C tip temperature (flow loss of -5.4, -5.06, -5.2, and -6.67%) indicated the test-to-test standard deviation of +/- 0.8% fuel flow loss or a coefficient of variation (COV) of 14%.

[0042] As is seen, increasing values of T₉₀ were found to be beneficial with respect to injector plugging, while higher olefin levels resulted in somewhat greater plugging tendencies. Sulfur was found to have a non-linear effect, producing a beneficial effect up to 150 ppm, and then reversing its effect up to 400 ppm.

[0043] To investigate the effect of aromatics, if any, on deposit formation a three fuel matrix was evaluated. This set of test matrix fuels is described in Table 3 below.

TABLE 3

	<u>Aro-1</u>	<u>Aro-2</u>	<u>Aro-3</u>
Component, vol. fct.			
Alkylate	70.0	30.0	--
FCC - Heavy	3.0	--	11.0
FCC - Light	--	--	21.0
FCC - Light	12.0	15.0	53.0
Reformate - 98 RON	15.0	55.0	15.0
Total, vol. fct.	100.0	100.0	100.0
Distillation (D86), F			
IBP	92	92	100
10 vol%	152	145	143
50 vol%	223	229	213
90 vol%	314	316	317
EP	413	406	415
RVP (D5191), psi	6.8	7.0	6.3
RON	93.5	95.6	90.7
MON	89.4	88.2	81.7
(R+M)/2	91.5	91.9	86.2
PIONA (M1530-14), vol%			
Saturates	81.4	58.7	50.7
Olefins	3.6	4.4	20.9
Aromatics	15.0	36.9	28.4
Gums (D381), mg/100 ml			
Unwashed	6	8	7
Washed	2	3	3
Sulfur (D5453), ppm	16	9	49
Density @ 60°F (D4052-1), g/cc	0.7192	0.7517	0.7459
Diene Number (M45), millimoles/g	0.3	0.1	0.4
Peroxide Number (M62), ppm	2.2	1.6	2.4

[0044] Aro-1 and Aro-2 were derived by adding the aromatics rich streams to
5 a base fuel stream of alkylate while Aro-3 is a mixture solely of heavy FCC,

light FCC, and reformat. In the case of Aro-2 the aromatics rich stream was predominately a reformat stream while for Aro-3 the aromatics stream was predominately a FCC stream.

- 5 [0045] Aro-1 was predominately alkylate with a minor quantity of a mixture of light FCC and reformat and some heavy FCC. The specifications for the heavy FCC, light FCC and reformat, 98 RON are presented in Table 4.

TABLE 4

10

	<u>Heavy FCC</u>	<u>Light FCC</u>	<u>Reformat</u>
IBP (°F)	246	99.5	95.2
T ₁₀	300	140	140
T ₉₀	406	290	320
FBP (°F)	453	337.8	399.4
RON	89.8	91	98.3
MON	79.7	80.7	88.2
Aromatics (vol%)	61.3	17.65	58.99
Olefins (vol%)	6.4	28.3	2.17
Benzene (vol%)	0.74	1.36	2.36
Sulfur, ppm	165	140	0

15

- [0046] These fuels were similarly tested in the 2.2 liter engine test bed rig previously described, at the same test conditions. The performance of each fuel in terms of injector tip deposit fouling as a function of tip temperature is presented in Figure 2.

20

- [0047] As is seen, Aro-2 causes somewhat lower tip deposit formation as evidenced by a lower percent flow loss as compared against Aro-1 (at temperature of 173°C) evidencing a minor but observable beneficial effect of increasing aromatics content of the fuel. Aro-3 similarly exhibited a reduction in the percent of flow loss but not as significant as that shown by Aro-2.

5

10

15

20

25

TABLE 5

SUMMARY	Regression Statistics
Multiple R	0.961144188
R Square	0.923798149
Adjusted R Square	0.885697224
Standard Error	0.417108099
Observations	7

ANOVA	df	SS	MS	F	Significance F
Regression	2	8.43662619	4.218313095	24.24608174	0.005806722
Residual	4	0.695916667	0.173979167		
Total	6	9.132542857			

	Coefficients	Standard Error	t Stat	P-value	Lower 95%	Upper 95%
Intercept	-8.416904762	0.236165056	-35.63992453	3.69937E-06	-9.072605433	
Heavy FCC	-0.76731746	0.110353451	-6.953271088	0.002247828	-1.073708394	
Light FCC	0.112126984	0.017410216	6.440298179	0.002990604	0.063788374	

	Coefficients	Standard Error	t Stat	P-value	Lower 95%	Upper 95%
Intercept	-10.53699405	0.403370644	-26.12236219	1.27606E-05	-11.65693282	-9.417055279
Total FCC	0.011619048	0.005439512	2.136045855	0.099529048	-0.003483491	0.026721586
Reformate	0.065958333	0.009519146	6.929017598	0.002277424	0.039528891	0.092387776

[0053] As is seen, the coefficient for "total FCC" is 0.0116 which is six times lower than the coefficient for reformat, indicating that of the two, the reformat is preferred as it will produce less injector flow loss. In each case the "P-value" for these coefficients was less than 0.1 indicating well over 90% statistical confidence, as does the adjusted "R squared" for the regression analysis of 88.5%.

[0054] Turning attention to heavy FCC versus light FCC, it is seen by reference to the coefficient values that heavy FCC has a coefficient which is negative (-0.767) which is more than seven times lower (detrimental) than the coefficient for light FCC (0.112).

[0055] Finally, Table 6 reports the overall fuel compositional profile for Aro-1, Aro-2 and Aro-3 and the result of seven runs in terms of GDI injector flow. Clearly the fuel containing the most reformat and the least heavy FCC/total FCC (Aro-2) resulted in the least negative impact on GDI injector flow rate. Aro-3 while containing the most total FCC also contained the most light FCC and comparing Aro-3 against Aro-1 it is seen that the fuel containing more light FCC (Aro-3) is superior in terms of GDI injector flow loss as compared against Aro-1 despite the fact that Aro-3 also contained more heavy FCC than did Aro-1.

103230"59275660

TABLE 6

	Olefins	Aromatic	Alky	Heavy FCC	Reformate	Light FCC	Total FCC	T90	Sulfur	End Point	GDI Flow
Aro-1	3.6	15	70	3	15	12	15	314	16	413	-8.89
	3.6	15	70	3	15	12	15	314	16	413	-9.57
	3.6	15	70	3	15	12	15	314	16	413	-9.66
Aro-2	4.4	36.9	30	0	55	15	15	316	9	406	-7.12
	4.4	36.9	30	0	55	15	15	316	9	406	-6.35
Aro-3	20.9	28.4	0	11	15	74	85	317	49	415	-8.41
	20.9	28.4	0	11	15	74	85	317	49	415	-8.71